Novel Coupling Reactions of Dithioacetals with Organocuprate Reagents. Propargylic Dithioacetal as an Allene-1,3-Zwitterion Synthon

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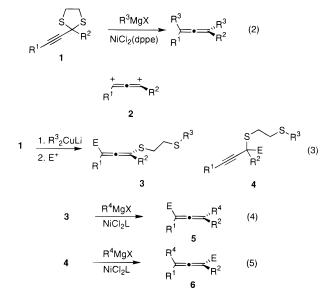
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The carbon-sulfur bond is known to be ambiphilic toward nucleophiles because the electronegativities for carbon and sulfur atoms are similar. Accordingly, different kinds of reagents or conditions may alter the selectivity of the reaction. It is well documented that the nickel-catalyzed cross-coupling reactions of organosulfur compounds with Grignard reagents provide a useful entry to convert a carbon-sulfur bond into a carbon-carbon bond.^{1,2} The carbon-sulfur bond in these reactions can be considered as a carbocation synthon. On the other hand, the production of a carbanionic leaving group from the corresponding thioether is rare unless a stabilized anionic species is produced. For example, the reaction of an (a-thioalkoxy)carbonyl compound with methyl mercaptide affords the corresponding desulfurized product.³ More recently, treatment of benzylic dithioacetals with organolithium reagents has led to the formation of the corresponding carbanions (eq 1).⁴

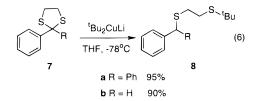
$$\begin{array}{c} SR^{1} & SR^{1} & SR^{1} \\ Ph-C-SR^{1} & \xrightarrow{RLi} & Ph-C-Li & \xrightarrow{E^{+}} & Ph-C-E & (1) \\ Me & Me & Me & Me \end{array}$$

Whereas propargylic acetals react readily with organocopper reagents to give the corresponding ring-opening products,⁵ the corresponding reaction with sulfur analogs has not been explored. We recently reported that the propargylic dithioacetal 1 can serve as an allene-1,3dication synthon 2 upon treatment with the Grignard reagent in the presence of a nickel catalyst (eq 2).^{2,6} In this regard, the displacement of the two carbon-sulfur bonds by two carbon-carbon bonds can be achieved in one pot. However, the reaction would be more versatile if the two carbon-sulfur bonds in 1 could be substituted by different moieties. In this paper, we report an unprecedented organocopper-induced C-S bond-cleavage reaction of propargylic dithioacetals leading to the corresponding thioethers 3 or 4. The combination of this process with the nickel-catalyzed cross coupling reaction

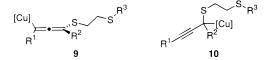


results in the sequential transformation of 1 into substituted allenes **5** or **6** (eqs 3-5).

In the beginning of this work, we tested the reactivity of benzylic dithioacetals 7 toward organocopper reagents, Thus, each of 7a and 7b was treated with 0.6 equiv of ^tBu₂CuLi at -78 °C for 30 min followed by quenching with MeOH to give 8a and 8b in 95 and 90% yields, respectively (eq 6). These promising preliminary results demonstrated the first examples of ring-opening reactions of dithiolanes with organocopper reagents.



An extension of this procedure to the reaction of 1a with the same copper reagent at -78 °C followed by quenching with MeOH at -78 °C gave the corresponding allenyl thioether 3a in 88% yield. When D_2O was employed to scavenge the reaction, deuterium incorporation at allenyl carbon (C₃) was observed (Table 1, entry 2). On the other hand, when 3 equiv of MeI and 2 equiv of HMPA⁷ were introduced at -78 °C instead of adding D_2O , methylation at the propargylic carbon occured to give 4a (Table 1, entry 7). Representative results are summarized in Table 1. Apparently, allenyl/propargyl organocopper intermediates 9 or 10 are involved in these reactions. It is noteworthy that neither 3 nor 4 reacted further with the organocopper reagent under these conditions.



As can be seen from Table 1, a number of allenyl/ parpogylic thioethers 3/4 can be easily obtained from the reactions of 1 with organocopper reagents. Sequential

⁽¹⁾ For reviews, see: (a) Naso, F. Pure Appl. Chem. 1988, 60, 79. (b) Fiandanese, V. Pure Appl. Chem. 1990, 62, 1987. (c) Luh, T.-Y.; Ni, Z.-J. Synthesis 1990, 89.

⁽²⁾ For reviews, see: (a) Luh, T.-Y. Acc. Chem. Res. 1991, 24, 257. (b) Luh, T.-Y.; Leung, M.-K. In Advances in the Use of Synthons in Organic Chemistry; Dondoni, A., Ed.; JAI: London, **1995**; Vol. 2, 129. (c) Luh, T.-Y. Pure Appl. Chem. 1996, 68, 105.

⁽³⁾ For example, see: Gassman, P. G.; Gilbert, D. P.; Luh, T.-Y. J. (4) Krief, A.; Kenda, B.; Barbeaux, P. *Tetrahedron Lett.* **1991**, *32*,

²⁵⁰⁹

^{(5) (}a) Normant, J. F.; Alexakis, A. *J. Organomet. Chem.* **1973**, *57*, C99. (b) Alexakis, A.; Mangeney, P.; Normant, J. F. *Tetrahedron Lett.* **1985**, *26*, 4197. (c) Alexakis, A.; Mangeney, P.; Ghribi, A.; Marek, I.; Sedrani, R.; Normant, J. F. *Pure Appl. Chem.* **1988**, 60, 49. (d) Alexakis, A.; Mangeney, P. *Tetrahedron: Asymmetry* **1990**, *1*, 477. (e) Marek, A., Mangeney, F. *Tetranearon: Asymmetry* **1990**, *1*, 477. (e) Marek, I.; Mangeney, P.; Alexakis, A.; Normant, J. F. *Tetrahedron Lett.* **1986**, *27*, 5499. (f) Alexakis, A.; Marek, I.; Mangeney, P.; Normant, J. F. *J. Am. Chem. Soc.* **1990**, *112*, 8042. (6) Tseng, H.-R.; Luh, T.-Y. J. Org. Chem. 1996, 61, 8685.

⁽⁷⁾ HMPA was essential for the alkylation with MeI or allyl bromide under these conditions (Table 1, entries 7 and 8). The yield was otherwise much lower. (Cf. Lipshutz, B. H. In *Organometallics in Synthesis*; Schlosser, M., Ed.; Wiley: Chichester, 1994; Chapter 4.)

entry	substrate	\mathbb{R}^1	R ²	R ³	E	product (% yield)
1	1a	Ph	Me	^t Bu	Н	3a (88)
2				^t Bu	D	3b (81)
3	1b	Ph	Ph	^t Bu	Н	3c (95)
4				ⁿ Bu	Н	3d (93)
5	1c	Bu	Me	^t Bu	Н	3e (82)
6	1d	TMS	Me	^t Bu	Н	4c (94)
7	1a	Ph	Me	^t Bu	Me	4a (81)
8				^t Bu	$CH_2 = CHCH_2$	4b (80) ^a

 $^{a}\,\mathrm{Crude}$ yield. The product was used directly for the next reaction.

 Table 2.
 NiCl₂(dppf)-Catalyzed Cross-Coupling Reactions of 2 or 6 with Grignard Reagents

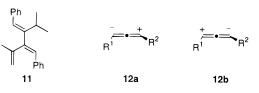
entry	substrate	R ⁴ MgX	product (% yield)
9	3a	MeMgI	5a (81)
10	3e	PhMgBr	5b (67)
11	3c	ⁱ PrMgBr ^a	5c (50)
12	4a	TMSCH ₂ MgCl	6a (72)
13	4b	ⁿ BuMgBr	6b (62)
14	4 c	PhMgBr	6c (67)
15	3a	MeMgI ^a	11 (52)

^a NiCl₂(dppe) was employed as the catalyst.

treatment of these thioethers with the Grignard reagent in the presence of 5 mol% of NiCl₂(dppf) resulted in a convenient synthesis of substituted allenes **5/6**. Table 2 summarizes representative examples. As expected,⁸ reduction of the C–S bond in **3c** was observed when ⁱ-PrMgBr was employed (Table 2, entry 11).

When NiCl₂(dppe) was employed as the catalyst in the coupling reaction of 3a, the corresponding dimeric prod-

uct **11** was obtained in 52% yield (Table 2, entry 15). The stereochemistry of **11** was determined by the NOE experiments. Dimerization of **5a** was also achieved in 59% yield upon treatment with MeMgI in the presence of NiCl₂(dppe).



In summary, we have demonstrated for the first time the use of propargylic dithioacetals as allene-1,3-zwitterion synthons **12**. By employing this procedure, di-, tri-, as well as tetra-substituted allenes are thus synthesized conveniently.

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Supporting Information Available: NMR spectra for **3a–e, 4a,c, 5b**, and **6a–c, 8a,b**, and **11** (14 pages).

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^{(8) (}a) Wenkert, E.; Hanna, J. M., Jr.; Leftin, M. H.; Michelotti, E. L.; Potts, K. T.; Usifer, D. J. Org. Chem. 1985, 50, 1125. (b) Wenkert, E.; Ferreira, T. W. J. Chem. Soc., Chem. Commun. 1982, 840. (c) Trost, B. M.; Lavoie A. C. J. Am. Chem. Soc. 1983, 105, 5075. (d) Trost, B. M.; Ornstein, P. L. J. Org. Chem. 1982 47, 748. (e) Shen, G.-Y.; Tapia, R.; Okamura, W. H. J. Am. Chem. Soc. 1987, 109, 7499. (f) Fabre, J.-L.; Julia, M. Tetrahedron Lett. 1983, 24, 4311. (g) Cuvigny, T.; Fabre, J.-L.; Hervé du Penhoat, C.; Julia, M. Tetrahedron Lett. 1983, 24, 4319. (h) Fabre, J.-L.; Julia, M.; Verpeaux, J.-L. Bull. Soc. Chim. Fr. 1985, 772. (i) Capet, M, Cuvigny, T.; Hervé du Penhoat, C.; Julia, M.; Loomis, G. Tetrahedron Lett. 1987, 28, 6273.